

## STEREOSELECTIVE TOTAL SYNTHESIS OF BAZZANENE

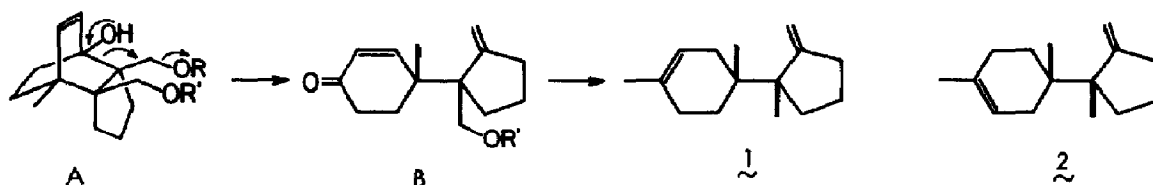
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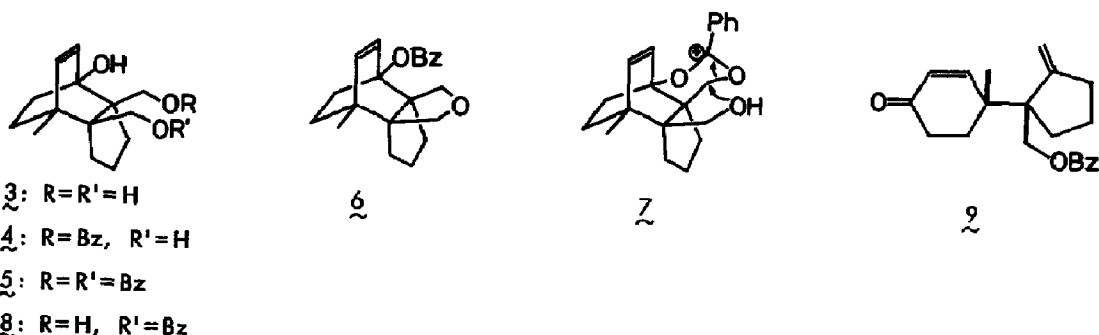
**Abstract** Bazzanene, the stereoisomer of trichodiene, isolated from a liverwort was synthesized with complete stereoselectivity in 8 steps (12% yield) starting from a known bicyclo[2.2.2]octane derivative

Bazzanene 1 is a bicyclic sesquiterpene isolated from a liverwort, *Bazzania pompeana*<sup>1)</sup>, and is the biogenetic precursor of the tricyclic barbatane (gymnomitrane) carbon skeleton. The first structure<sup>1)</sup> proposed was revised to the presently accepted structure 1<sup>2)</sup> on the basis of NMR and CD spectra and the chemical correlation with trichodiene, 2, a fungal metabolite<sup>3)</sup>. This paper describes a stereoselective total synthesis of bazzanene 1.



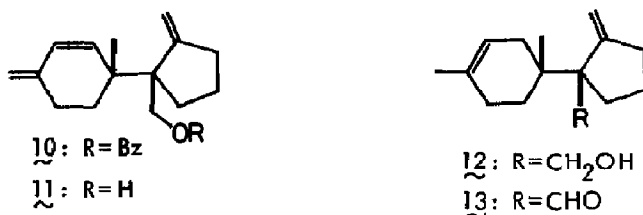
Our strategy involves the C-C bond cleavage (arrow) of a properly substituted tricyclo[5.2.2.0<sup>2,6</sup>]-undecane A with better leaving group OR than OR'. This single process (A → B) would not only create an exocyclic double bond on 5-membered ring, but also forms a cyclohexenone moiety in the product B, allowing the discrimination of  $\alpha$  and  $\alpha'$  positions of the carbonyl group, and leading stereoselectively to 1.

We have started the synthesis from the triol 3 which was one of the intermediates in our recent total synthesis of barbatane sesquiterpenes<sup>4)</sup>. Benzoylation of 3 afforded the monobenzoate 4, m.p. 132-133°, and the dibenzoate 5, viscous oil, in 25% and 59% yields, respectively. Although attempted protection primary hydroxyl group in 4 as tetrahydropyranyl ether furnished only the ether benzoate 6, m.p. 138.5-139° (probably via 7), mild hydrolysis of 5 with catalytic amount of NaOH in MeOH at room temperature afforded another benzoate 8, m.p. 111-111.5°, in 45% yield together with 3 (35%) and the recovered 5 (14%). Only a trace of 4 was detected<sup>5)</sup>. The 1,3-glycol structure in 8 was confirmed by the formation of an acetonide, m.p. 118-119.5° (CuSO<sub>4</sub>/Acetone).



When the mesylate obtained from  $\underline{8}$  by MsCl in pyridine was treated with KOtBu in THF at room temperature, the crucial bond cleavage took place cleanly and the expected cyclohexenone  $\underline{9}$  was obtained in 75% yield from  $\underline{8}$ .  $\underline{9}$ : m/e 324 ( $M^+$ ), 105 (b.p.),  $\nu$  1712, 1670, 885  $\text{cm}^{-1}$ ,  $\delta$  ( $\text{CDCl}_3$ ) 1.25 (3H, s), 4.38 (2H, s), 4.97 (1H, br.s), 5.21 (1H, br.s), 5.91 (1H, d,  $J=10$ ), 6.88 (1H, dd,  $J=10, 1.5$ ).

For the regioselective construction of methylcyclohexene moiety in  $\underline{1}$ ,  $\underline{2}$  was subjected to Wittig reaction ( $\text{CH}_2=\text{PPh}_3$  in THF) by which two products, the triene benzoate  $\underline{10}$ , oil, and the triene alcohol  $\underline{11}$ , oil, were obtained in 52% and 36% yields, respectively. Bouveault-Blanc reduction ( $\text{Na}/n\text{BuOH}$ , 8l of both  $\underline{10}$  and  $\underline{11}$ ) afforded the desired 1,4-reduction product  $\underline{12}$  in 93-94% yield. The conversion of the hydroxymethyl group in  $\underline{12}$  to a methyl group was achieved by  $\text{CrO}_3$  oxidation to the aldehyde  $\underline{13}$  and the subsequent Huang-Minlon reduction (70% overall yield). The hydrocarbon obtained exhibited the identical



MS, IR and NMR spectra with natural bazzanene<sup>6)</sup>. Thus, total synthesis of  $\underline{1}$  was achieved with complete stereoselectivity.

#### References and Notes

- 1) S. Hayashi, A. Matsuo and T. Matsuura, *Experientia*, **25**, 1139 (1969); A. Matsuo, *Tetrahedron*, **27**, 2757 (1971).
- 2) A. Matsuo and S. Hayashi, *J.C.S. Chem. Comm.*, 566 (1977).
- 3) S. Nozoe and Y. Machida, *Tetrahedron Letters*, 2761 (1970). *Idem*, *Tetrahedron*, **28**, 5105 (1972). Y. Machida and S. Nozoe, *ibid.*, **28**, 5113 (1972).
- 4) M. Kodama, T. Kurihara, J. Sasaki and S. Itô, *Can. J. Chem.*, **57**, 3343 (1979).
- 5) This selectivity in hydrolysis is apparently due to the participation of the neighboring tert-OH group [S. M. Kupchan, et al., *J. Am. Chem. Soc.*, **85**, 350 (1963), **88**, 343, 347 (1966)]. Once a benzo group was liberated, the hydrolysis of the second benzoyl group took place rather rapidly (again by the similar participation), so that the change of the hydrolytic condition had little effect on the yield of
- 6) The authors are deeply indebted to Professor S. Hayashi and Dr. A. Matsuo, Hiroshima University, for the spectra of bazzanene and to Professor S. Nozoe, Tohoku University, for the spectra of trichodien

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